Stat Phys 2009 Test 2 02-24-09

$$\int_{0}^{\infty} \exp(-ax^{2}) dx = \frac{\sqrt{\pi}}{2\sqrt{a}} \qquad \int_{0}^{\infty} x^{2} \exp(-ax^{2}) dx = \frac{\sqrt{\pi}}{4a^{3/2}}$$

Problem 1

Consider a zipper with infinite number of links. Each link can be in one of the two states: a state with energy 0 wherein the link is closed, and a state with energy $\varepsilon > 0$ wherein it is open. This zipper can unzip only from one end and a link can open only if all the links before it (from that end) are already open. At temperature *T*, find the average number of open links and investigate your answer for high and low temperatures.

Solution

Partition function is

$$Z = \sum_{n=0}^{\infty} \exp\left(-\frac{n\varepsilon}{T}\right) = \frac{1}{1 - \exp\left(-\frac{\varepsilon}{T}\right)}$$

whereof

$$\begin{split} \langle n \rangle &= \frac{1}{Z} \sum_{n=0}^{\infty} n \exp\left(-\frac{n\varepsilon}{T}\right) = -\frac{T}{Z} \frac{\partial}{\partial \varepsilon} \sum_{n=0}^{\infty} \exp\left(-\frac{n\varepsilon}{T}\right) = -\frac{T}{Z} \frac{\partial}{\partial \varepsilon} Z = \frac{1}{\exp\left(\frac{\varepsilon}{T}\right) - 1} \\ \langle n \rangle &\approx \exp\left(-\frac{\varepsilon}{T}\right), \qquad \text{for } T <<\varepsilon \\ \langle n \rangle &\approx \frac{T}{\varepsilon}, \qquad \qquad \text{for } T >> \varepsilon \end{split}$$

Alternatively, energy (average energy) of the system is

$$E = -\frac{\partial \log Z}{\partial \theta} = \frac{\partial}{\partial \theta} \log \left[1 - \exp(-\theta \varepsilon)\right] = \frac{\varepsilon}{\exp(\theta \varepsilon) - 1} = \frac{\varepsilon}{\exp\left(\frac{\varepsilon}{T}\right) - 1}$$

and

$$\langle n \rangle = \frac{E}{\varepsilon}$$

Problem 2

A classical system of N distinguishable (no N!), non-interacting particles is placed in a three-dimensional harmonic well

$$U(r) = \frac{x^2 + y^2 + z^2}{2a^2}$$

Find the Helmholtz free energy and the internal energy of the gas.

Solution

Partition function for each particle is found as

$$Z_{1} = \frac{1}{\left(2\pi\hbar\right)^{3}} \int_{0}^{\infty} \exp\left(-\frac{p^{2}}{2mT}\right) 4\pi p^{2} dp \int_{-\infty}^{\infty} \exp\left(-\frac{x^{2}+y^{2}+z^{2}}{2a^{2}T}\right) dx dy dz = \frac{T^{3}a^{3}m^{3/2}}{\hbar^{3}}$$

 $F = -TN \log Z_1$

$$E = -N\frac{\partial \log Z_1}{\partial \theta} = N\frac{\partial}{\partial \theta}\log\frac{\hbar^3\theta^3}{a^3m^{3/2}} = \frac{3N}{\theta} = 3NT$$

Problem 3 (bonus)

An ideal gas, PV = NT, with the initial pressure P_0 , and volume V_0 is heated by a current through a platinum wire. The experiment is done twice: first at a constant volume V_0 , with pressure changing from P_0 to P_1 , and then at a constant pressure P_0 , with volume changing from V_0 to V_1 . In both instances, gas receives the same amount of heat (current is applied for the same amount of time). Assuming constant specific heat, find C_P/C_V .

Solution

For constant volume,

$$dQ = dE = C_V dT$$

$$\Delta Q = C_V \Delta T = C_V (T_1 - T_0)$$

For constant pressure,

$$dQ = dW = C_P dT$$

$$\Delta Q = C_P \Delta T = C_P (T_2 - T_0)$$

whereof

$$\frac{C_P}{C_V} = \frac{T_1/T_0 - 1}{T_2/T_0 - 1} = \frac{P_1/P_0 - 1}{V_1/V_0 - 1}$$