Stat Phys 2009 Test 2 02-24-09

$$
\int_0^\infty \exp(-ax^2)dx = \frac{\sqrt{\pi}}{2\sqrt{a}} \qquad \int_0^\infty x^2 \exp(-ax^2)dx = \frac{\sqrt{\pi}}{4a^{3/2}}
$$

Problem 1

Consider a zipper with infinite number of links. Each link can be in one of the two states: a state with energy 0 wherein the link is closed, and a state with energy $\varepsilon > 0$ wherein it $\frac{1}{2}$ is open. This zipper can unzip only from one end and a link can open only if all the links before it (from that end) are already open. At temperature *T* , find the average number of open links and investigate your answer for high and low temperatures.

Solution

Partition function is

$$
Z = \sum_{n=0}^{\infty} \exp\left(-\frac{n\varepsilon}{T}\right) = \frac{1}{1 - \exp\left(-\frac{\varepsilon}{T}\right)}
$$

whereof

$$
\langle n \rangle = \frac{1}{Z} \sum_{n=0}^{\infty} n \exp\left(-\frac{n\varepsilon}{T}\right) = -\frac{T}{Z} \frac{\partial}{\partial \varepsilon} \sum_{n=0}^{\infty} \exp\left(-\frac{n\varepsilon}{T}\right) = -\frac{T}{Z} \frac{\partial}{\partial \varepsilon} Z = \frac{1}{\exp\left(\frac{\varepsilon}{T}\right) - 1}
$$

$$
\langle n \rangle \approx \exp\left(-\frac{\varepsilon}{T}\right), \qquad \text{for } T < \varepsilon
$$

$$
\langle n \rangle \approx \frac{T}{\varepsilon}, \qquad \text{for } T > \varepsilon
$$

Alternatively, energy (average energy) of the system is

$$
E = -\frac{\partial \log Z}{\partial \theta} = \frac{\partial}{\partial \theta} \log[1 - \exp(-\theta \varepsilon)] = \frac{\varepsilon}{\exp(\theta \varepsilon) - 1} = \frac{\varepsilon}{\exp(\frac{\varepsilon}{T}) - 1}
$$

and

$$
\langle n \rangle = \frac{E}{\varepsilon}
$$

Problem 2

A classical system of *N* distinguishable (no *N*!), non-interacting particles is placed in a three-dimensional harmonic well

$$
U(r) = \frac{x^2 + y^2 + z^2}{2a^2}
$$

Find the Helmholtz free energy and the internal energy of the gas.

Solution

Partition function for each particle is found as

$$
Z_{1} = \frac{1}{(2\pi\hbar)^{3}} \int_{0}^{\infty} \exp\left(-\frac{p^{2}}{2mT}\right) 4\pi p^{2} dp \int_{-\infty}^{\infty} \exp\left(-\frac{x^{2} + y^{2} + z^{2}}{2a^{2}T}\right) dx dy dz = \frac{T^{3} a^{3} m^{3/2}}{\hbar^{3}}
$$

 $F = -TN \log Z_1$

$$
E = -N \frac{\partial \log Z_1}{\partial \theta} = N \frac{\partial}{\partial \theta} \log \frac{\hbar^3 \theta^3}{a^3 m^{3/2}} = \frac{3N}{\theta} = 3NT
$$

Problem 3 (*bonus*)

An ideal gas, $PV = NT$, with the initial pressure P_0 , and volume V_0 is heated by a current If pressure enanging from T_0 to T_1 , and then at a const through a platinum wire. The experiment is done twice: first at a constant volume V_0 , $\ddot{}$ with pressure changing from P_0 to P_1 , and then at a constant pressure P_0 , with volume who and of the p. Thoughing constant of changing from V_0 to V_1 . In both instances, gas receives the same amount of heat (current is applied for the same amount of time). Assuming constant specific heat, find C_p / C_v .

€ *Solution*

For constant volume,

$$
dQ = dE = C_V dT
$$

$$
\Delta Q = C_V \Delta T = C_V (T_1 - T_0)
$$

For constant pressure,

$$
dQ = dW = C_p dT
$$

$$
\Delta Q = C_p \Delta T = C_p (T_2 - T_0)
$$

whereof

$$
\frac{C_P}{C_V} = \frac{T_1/T_0 - 1}{T_2/T_0 - 1} = \frac{P_1/P_0 - 1}{V_1/V_0 - 1}
$$